## Chem 1010/1800 Tip Sheet

## Significant Figures

Scientists use significant figures (s.f.) to track the precision of their measurements. When doing calculations, we can never end up with a result that is more precise than what we started with. This means we have to follow rules to ensure we use the correct number of significant figures.

## How to do it

1. Any non-zero digits are significant
2. Any zero between two non-zero digits is significant
3. Leading zeroes are not significant
4. Zeroes that appear after a decimal are significant, unless they are leading
5. Exact numbers have an infinite number of significant figures

## Examples

Counting Significant
Figures

Addition \& Subtraction

- Count the number of decimals in every term being added or subtracted. Determine which term has the fewest number of decimals
- Compute the answer, then round to the fewest number of decimals from the original terms

1. $125 \mathrm{~mL} . . .3$ s.f.
2. $305 \mathrm{~g} . . .3$ s.f.
3. $0.013 \mathrm{~L} . . .2$ s.f.
4. $20.00 \mathrm{~atm} . . .4$ s.f.
5. 9 candies ... exact number (infinite s.f.)

- $1.21+0 . \underline{5}=1.7$
- $781 . \underline{6}-224=558$
- $0.004432-0.1938+$ $1.00399=0.8146$
- $1.21 \times 0 . \underline{5}=0 . \underline{6}$
- $\underline{781.6} \div \underline{224}=\underline{3.49}$
- $0.004432 \times 0.1938 \div$ $1.00399=8.555 \times 10^{-4}$
- Count the number of significant figures $-\log (\underline{12})=1.08$ inside the logarithm
Logarithms
- Compute the answer, then round the number of decimal places to the number of sig figs from inside the logarithm
- $\log (0 . \underline{5008})=-0 . \underline{3003}$
- $\ln ($ ®2 1$)=4145$
- Count the number of decimal places in the exponent
Exponents
- Compute the answer, then round the number of sig figs to the number of decimals from the exponent
- $2^{1.40}=\underline{2.6}$
- $100 . \underline{9795}=\underline{9.539}$
- $0.5^{7.1}=0.00 \underline{5}$


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